



## Concepts in Toxicology

Principles of Environmental Toxicology  
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Principles of Environmental Toxicology

### Learning Objectives

- Define toxicology and toxicity.
- Discuss different types of toxic responses.
- Explain how toxicants are classified.
- Describe the phases of toxicosis.
- Explain how concomitant exposure influences toxicity.
- Develop an introductory understanding of toxicity testing.

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## Toxicology

- The science that deal with the adverse effects of chemicals on living systems.
- Classifications.
  - Descriptive toxicology.
    - What?
  - Mechanistic toxicology.
    - Why?
  - Analytical toxicology.
    - How much?



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Principles of Environmental Toxicology

### Definition of Toxicity

- Toxicity: The degree to which a substance can harm humans or animals.
- Toxicity can be acute, subchronic, or chronic.

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## Acute Toxicity

- Involves harmful effects in an organism through a single or short-term exposure.



The Death of Socrates, 1787 Jacques-Louis David  
(Metropolitan Museum of Art, New York)

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Principles of Environmental Toxicology

### Subchronic Toxicity

- The ability of a toxic substance to cause effects for more than one year but less than the lifetime of the exposed organism.



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## Chronic Toxicity

- The ability of a substance or mixture of substances to cause harmful effects over an extended period, usually upon repeated or continuous exposure, sometimes lasting for the entire life of the exposed organism.



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## Specialty Areas in Toxicology

- Target Organ/System.**
  - Neurotoxicology, Genetic Toxicology, Reproductive Toxicology, Immunotoxicology, Endocrine Toxicology.
- Target Species/Systems.**
  - Aquatic Toxicology, Environmental Toxicology, Wildlife Toxicology, Veterinary Toxicology.
- Selected Responses.**
  - Teratology, Carcinogenesis.

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## Applied Toxicology

- Occupational toxicology.
- Clinical toxicology.
  - Toxic induced diseases and antidotes.
- Forensic toxicology.
  - Determining causes of death.
- Regulatory toxicology.
  - Risk assessment from descriptive tests
- Developmental toxicology.
  - New chemicals and uses.

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## Classification of Toxicants

- Target organ.**
  - Hepatotoxin, neurotoxin.
- Intended use.**
  - Pesticide, solvent.
- Source.**
  - Natural, synthetic.
- Special effect.**
  - Carcinogen, mutagen, endocrine disruptor.



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## Classification of Toxicants, 2

- Physical state.
  - Gas, solid.
- Toxicity.
  - Extremely, slightly.
- Chemical composition.
  - Heavy metal, organophosphate.
- Mechanism of action.
  - Anticholinergic, inhibitor, uncoupler.



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## Types of Toxic Responses

- Local.**
  - Effect at site of contact.
  - GIT, lungs.
- Systemic.**
  - Effect distant from exposure site.
  - CNS, kidney, lungs.
- Some both.**
- Immediate.**
  - Minutes to hours after a single exposure.
- Delayed.**
  - Days to years after exposure.
- Some both.**

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## Types of Toxic Responses, 2

- Reversible vs. Irreversible
- Largely determined by
  - Tissue involved, length of exposure and magnitude of toxic insult.
- Reversible - rapidly regenerating tissue.
  - Liver, intestinal mucosa, blood cells.
- Irreversible
  - CNS damage, carcinogenesis, mutagenesis, teratogenesis.



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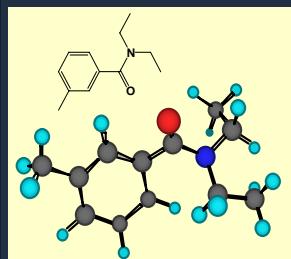
## Bioavailability

- Octanol-Water Partition Coefficient,  $K_{ow}$
- An empirical solubility term that can be used to assess transmembrane movement potential.
- $K_{ow} = 10^2$  to  $10^3$  indicates good chemical for absorption ( $\log K_{ow} = 2$  to 3).
  - OK lipid solubility and OK water solubility.

$$K_{ow} = [T]_{\text{octanol}} / [T]_{\text{water}}$$

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## N,N-diethyl-3-methylbenzamide



Experimental  $\log K_{ow} = 2.18$

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## Three Phases of Toxicology

- Exposure phase.
- Toxicokinetic phase.
  - Absorption.
  - Distribution.
  - Metabolism.
  - Excretion.
- Toxicodynamic phase.



## Exposure Phase

- Bioavailability.
  - The fraction of a dose available for absorption.
- Main factors.
  - Time and frequency of exposure, e.g. acute, subchronic...
  - Route of administration.
    - Animal: oral, lung, skin, injection.
    - Plant: roots, leaves.

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## Exposure Phase, 2



- Dose.
- Physical and chemical form of the toxicant.
  - Particle size, solubilization.

- Host related factors.
- Pre-absorption metabolism.

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## Absorption Phase

- Comparative aspects.
  - Cellular to organism.
- Membrane morphology.
  - Lipoprotein bilayer.
- Physicochemical processes that govern transmembrane movement.
  - Lipid-water solubility,  $K_{ow}$
  - Ionization ( $pK_a$ ), functional groups
  - Molecular size and conformation.

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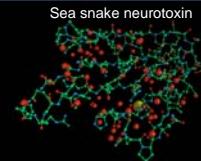
## Absorption Phase, 2

- Transmembrane movement.
  - Simple diffusion – Fick's Law.
  - Filtration – aqueous pores.
  - Carrier mediated.
- Sites of Absorption.
  - Animals – GIT, dermal, lung.
  - Plants – stomatal pores, cuticle, roots.
  - Insects – pore canals, oral.
  - Fish – gills, GIT, dermal.

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## Distribution Phase

- Four fates.
  - Site of toxic action, storage, metabolism, excretion.
- How it occurs.
  - Animals – blood, lymph.
  - Plants – xylem and/or phloem.
- Barriers of toxicological significance.
  - Blood/brain.
  - Placental (maternal - fetal).
  - Mammary (blood - milk).



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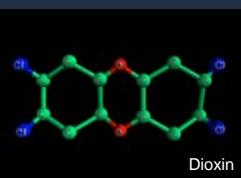
## Distribution Phase, 2

- Factors affecting distribution.
  - Affinity of tissues for the xenobiotic.
  - Blood flow, protein binding.
  - Route of administration, rate of metabolism.
- Redistribution.
  - Enterohepatic recirculation.

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## Metabolism Phase

- Phase I – Bioconversion.
  - Factors affecting toxicity and metabolism.
    - Environmental, genetic...
- Phase II – Conjugation.
  - "Grease to salt"



Dioxin

## Factors Influencing Toxicity

### Concomitant Exposure

- Additive       $2 + 2 = 4$ 
  - 2 OP's leading to cholinesterase inhibition.
- Synergistic       $2 + 2 = 10$ 
  - $CCl_4$  with ethanol leading to hepatotoxicity.
- Potentiation       $2 + 0 = 6$ 
  - Isopropanol with  $CCl_4$ , tumor promoters.
- Antagonism       $2 + 2 = 0$ 
  - BAL with heavy metals, antidotes.

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## Excretion

- Toxicological significance.
- Renal excretion.
- Non-renal excretion.
  - Biliary, expiration, gastric secretion...
- Comparative aspects.
  - Animals, plants.



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## Toxicodynamics

### Dose - Response Relationships



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## Intrinsic Activity

- Intrinsic activity: response
  - Agonist - substances with intrinsic activity, e.g.  $O_2$
  - Antagonist - substances that work against agonist, e.g. CO



Drugs – Typically reversible.  
Toxicants – Typically non-reversible.

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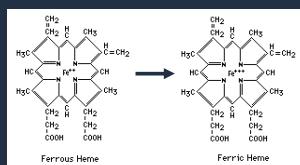
## Oxygen Transport Toxicants

- Methemaglobin formation ( $Fe^{2+}$  to  $Fe^{3+}$ ).
  - Nitrate, nitrite.
  - Naphthalene.
  - Chlorate.
  - Acetaminophen.
- $O_2$  competition at  $Fe^{2+}$ 
  - CO, carbon monoxide.
  - $CN^-$ , cyanide.



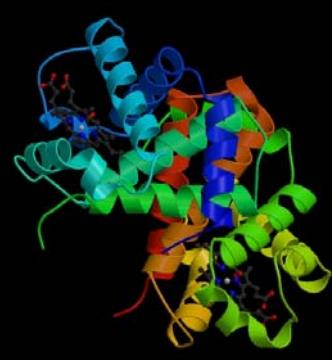
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## Methemaglobin Formation



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## $O_2$ Transport, Hemoglobin



Bar-headed goose  
Hemoglobin  
(oxy form)

## Toxicity Rating – Oral Human Dose

Class	Dose	For Average Adult
Practically non-toxic	>15 g/kg	More than a quart
Slightly toxic	5-15 g/kg	Between a pint and a quart
Moderately toxic	0.5-5 g/kg	Between an ounce and a pint
Very toxic	50-500 mg/kg	Between a teaspoonful and an ounce
Extremely toxic	5-50 mg/kg	Between 7 drops and a teaspoonful
Supertoxic	<5 mg/kg	A taste (<7 drops)

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## Spectrum of Toxic Dose

Agent	LD <sub>50</sub> (mg/kg)	Fish berry
Ethanol	10,000	
NaCl	4,000	
Ferrous sulfate	1,500	
Morphine sulfate	900	
Phenobarbital	150	
DDT	100	
Picrotoxin	5	
Strychnine sulfate	2	
Nicotine	1	
d-Tubocurarine	0.5	
Tetrodotoxin	0.1	
Dioxin (TCDD)	0.001	
Botulinus toxin	0.00001	

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